33. 1, 1-Dihalogenoanthra [b]cyclopropenes

Preliminary communication¹)

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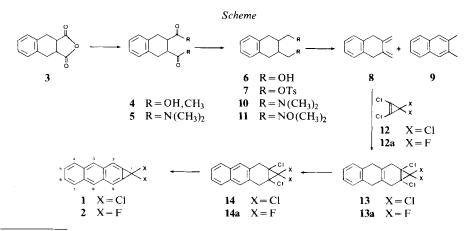
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Summary

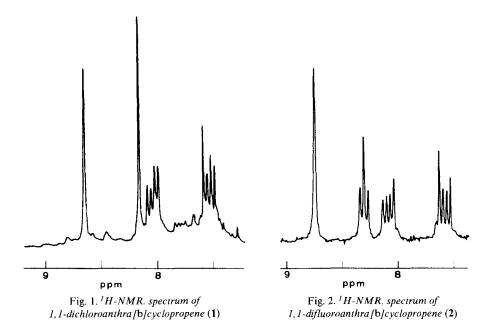
1, 1-Dichloro- and 1, 1-difluoroanthra[b]cyclopropene (1 and 2) are synthesized in 8 steps from the *Diels-Alder* adduct of benzocyclobutene and maleic anhydride.

Although the chemistry of the highly strained benzo- [1] and naphthocyclopropenes [2] is well established, the benzannelated homologues have so far remained elusive. To our knowledge, the only reports deal with the attempted synthesis of anthra[b]cyclopropene via the Billups [3] route. We describe the synthesis of the dichloro- and difluoroanthra[b]cyclopropenes 1 and 2 by a straightforward extension of our reported sequence for preparation of 1,1-dihalogenonaphtho[b]cyclopropenes [4]. The Diels-Alder adduct 3 [5] of benzocyclobutene [6] and maleic anhydride (Scheme) was hydrolyzed in boiling methanol to the diester 4. Reduction with LiAlH₄ afforded the diol 6 which was converted to the ditosylate 7. Unfortunately, elimination with tBuOK in THF did not give the desired diene 8, but almost exclusively 2,3-dimethylnaphthalene (9). Although a variety of elimination procedures were tried, the base-catalyzed



¹⁾ The full paper will be published in Helv.

aromatization to 9 could not be suppressed. An approach via thermal elimination was tried, using the methodology described for the preparation of 4,5-dimethylidenecyclohexene [7]. The diacid 4 was converted to the diamide 5 with hexamethylphosphorous amide in 95% yield. Reduction with LiAlH₄ gave the diamine 10 (90%) which was oxidized to the N-oxide 11. The latter was pyrolyzed without purification at $180^{\circ}/10^{-2}$ Torr to give 1,2,3,4-tetrahydro-2,3-dimethylidenenaphthalene (8) [20-30%; ¹H-NMR.: 7.0 (s), 5.26 and 4.83 (m, vinylic H), 3.45 (m, aliphatic H)]. Diels-Alder addition of 8 to tetrachlorocyclopropene (12) in presence of $NaHCO_3$ proceeded at RT. in 40 h (62%). Similarly, with 1,2-dichloro-3,3-difluorocyclopropene (12a) [8] the adduct 13a was formed in 22% yield after 6 days at 45°. Dehydrohalogenation to 1 and 2 was carried out with 2.2 mol-equiv. of *i*BuOK in THF at -70° to 25°, followed by anhydrous work-up. 1, 1-Dichloroanthra[b]cyclopropene (1) was obtained in ca. 70% yield together with some unreacted 13. The compound showed a strong tendency to decompose, and could not be isolated pure. However, its structure could be deduced unambiguously from its ¹H-NMR. spectrum (Fig. 1) and from the spectrum of the difluoro derivative 2 (Fig. 2). The latter compound, which was formed in 88% yield had m.p. 192-194° (from CHCl₂).



The MS. showed main peaks at m/z 226 (100, M^+); 207 (16, $M^+ - F$); 176 (19, $M^+ - CF_2$). The ¹H-NMR. spectrum (*Fig. 2*) displays the expected pattern with the AA'BB'-system of the H-atoms at C(4)-C(7); the triplet at 8.31 ppm is due to H-C(2, 9) (J(H,F)=3.75 Hz) and the singlet at 8.75 to H-C(3, 8). In the ¹⁹F-NMR. spectrum the corresponding triplet is found at 77.6 ppm downfield from C₆F₆. The ionization of dihalogenoanthra[b]cyclopropenes to form aromatic cations and the reduction to unsubstituted hydrocarbons is currently under investigation.

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REFERENCES

- [1] B. Halton, Chem. Rev. 73, 113 (1973).
- [2] W.E. Billups, Acc. Chem. Res. 11, 245 (1978).
- [3] D. Davalian & P.J. Garratt, Tetrahedron Lett. 1976, 2815; W. E. Billups & M. Asim, quoted in [2].
- [4] P. Müller & H. C. Nguyen Thi, Tetrahedron Lett. 21, 2145 (1980).
- [5] K. Alder & M. Fremery, Tetrahedron 14, 190 (1961).
- [6] A. Sanders & W. P. Giering, J. Org. Chem. 38, 3055 (1973); M. P. Cava & D. R. Napier, J. Am. Chem. Soc. 79, 1701 (1957).
- [7] E. D. Middlemas & L. D. Quin, J. Org. Chem. 44, 2587 (1979).
- [8] J. Sepiol & R. L. Soulen, J. Org. Chem. 40, 3791 (1975).